

Ag-Al-Gd (Silver-Aluminum-Gadolinium)

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Recently, [2004Ste] determined an isothermal section for this ternary system at 497 °C for Gd content up to 50 at.%. Seven ternary phases are present at this temperature.

(Zr₃Al₂-type tetragonal), and Gd₂Al (C23, Co₂Si-type orthorhombic). See [Massalski2] for the above binary diagrams.

Binary Systems

The Ag-Al phase diagram depicts three intermediate phases: Ag₃Al (20.5-29.8 at.% Al; denoted β, body-centered cubic; stable between 778 and ~600 °C), Ag₂Al (22.9-41.9 at.% Al; denoted δ, close-packed hexagonal), and μ (21-24 at.% Al, stable below 450 °C; A13, βMn-type cubic). In the Ag-Gd system, the intermediate phases are: Gd₁₄Ag₅₁ (Gd₁₄Ag₅₁-type tetragonal), GdAg₂ (C11_b, MoSi₂-type tetragonal), and GdAg (B2, CsCl-type cubic). The Al-Gd system has the following intermediate phases GdAl₃ (D0₁₉, Ni₃Sn-type hexagonal), GdAl₂ (C15, MgCu₂-type cubic), GdAl (ErAl-type orthorhombic), Gd₃Al₂

Ternary Isothermal Section

With starting metals of 99.95% Ag, 99.99% Al, and 99.5% Gd, [2004Ste] arc-melted under Ar atm 60 alloys with Gd content up to 50 at.%. The samples were annealed at 497 °C (770 K) for 1000 h and quenched in water. The phase equilibria were studied by x-ray powder diffraction. The isothermal section at 497 °C (770 K) constructed by [2004Ste] is shown in Fig. 1, to agree with the accepted binary data. The structural details of the seven ternary compounds (denoted τ₁ to τ₇ here and as 1 to 7 by [2004Ste]) are listed in Table 1. The compounds τ₁, τ₄, and τ₅ show a significant homogeneity range. The other ternary compounds

Table 1 Ag-Al-Gd crystal structure and lattice parameter data [2004Ste]

Phase	Composition, at.%	Pearson symbol	Space group	Prototype	Lattice parameter, nm
Gd _{1.85} Ag _{8.2-9.4} Al _{8.8-7.6} (τ ₁)	43.5-49.9 Ag 46.7-40.3 Al 9.8 Gd	<i>hP</i> 38	<i>P</i> 6 ₃ / <i>mmc</i>	Th ₂ Ni ₁₇	<i>a</i> = 0.93026* <i>c</i> = 0.90991
Gd ₈ Ag ₂₀ Al ₄₆ (τ ₂)	27 Ag 62.2 Al 10.8 Gd	<i>tP</i> ?	<i>I</i> 4/ <i>mmm</i>	Yb ₈ Cu ₁₇ Al ₄₉	<i>a</i> = 0.88177 <i>c</i> = 1.68624
~GdAg _{3.5} Al _{1.5} (τ ₃)	58.3 Ag 25 Al 16.7 Gd	<i>h</i> ??	Hex.	...	<i>a</i> = 0.5400 <i>c</i> = 0.9268
GdAg _{2.2-2.8} Al _{2.8-2.2} (τ ₄)	36.7-46.7 Ag 46.7-36.7 Al 16.7 Gd	<i>hP</i> ?	<i>P</i> 6 ₃ / <i>mmc</i>	DyAg _{2.4} Al _{2.6}	<i>a</i> = 0.92305* <i>c</i> = 0.94111
Gd ₃ Ag _{2.0-2.7} Al _{9.0-8.3} (τ ₅)	14.3-19.3 Ag 64.3-59.3 Al 21.4 Gd	<i>oI</i> 28	<i>I</i> <i>mmm</i>	αLa ₃ Al ₁₁	<i>a</i> = 0.43306* <i>b</i> = 1.26706 <i>c</i> = 1.00145
Gd ₃ Ag ₂ Al ₇ (τ ₆)	16.7 Ag 58.3 Al 25 Gd	<i>hR</i> 12	<i>R</i> $\bar{3}m$	Ca ₃ Cu ₂ Al ₇	<i>a</i> = 0.55668 <i>c</i> = 2.6382
GdAg _{0.8} Al _{1.2} (τ ₇)	26.7 Ag 40 Al 33.3 Gd	<i>oI</i> 12	<i>I</i> <i>mma</i>	CeCu ₂	<i>a</i> = 0.45899 <i>b</i> = 0.72479 <i>c</i> = 0.7889
GdAg _{5.2} Al _{6.8}	40 Ag 52.3 Al 7.7 Gd	<i>tI</i> 26	<i>I</i> 4/ <i>mmm</i>	ThMn ₁₂	<i>a</i> = 0.91555 <i>c</i> = 0.54318

* Lattice parameters are at Gd_{1.85}Ag_{9.3}Al_{7.7}, GdAg_{2.3}Al_{2.7}, and Gd₃Ag_{2.55}Al_{8.45} respectively

